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Standardized Practice Test Chemistry Solutions

Chemistry Concepts and Applications Chapter 4: Formation of Compounds Standardized Test Practice

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8. Which of the following acids could be the unknown solution given that the unknown solution had an initial volume of 12.5 mL and a concentration of 0.10 M? F. HCl G. HNO₃ H. HC₂H₃O₂ J. H₂SO₄ 9. Which concentration indicates a basic solution at 298 K? A. [OH⁻] > 1.0 × 10⁻⁷ C. [H⁺] > 1.0 × 10⁻⁷ B. [OH⁻] = 1.0 × 10⁻⁷ D. [H⁺] = 1.0 × 10⁻⁷

Chapter 16 Standardized Test Practice

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Chapter 15 Standardized Test Practice 1. Saturated solutions can be: A. concentrated only. C. either dilute or concentrated. B. dilute only. D. neither concentrated nor dilute. 2. Molarity is defined as the number of moles of: F. solute per liter of solvent. H. solution per liter of solvent. G. solute per liter of solution. J. solvent per liter ...

Chapter 15 Standardized Test Practice

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