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standard temperature

& pressure (STP)

Definition of Dalton's

law of partial

pressures. Definitions

& formulas for Boyle's,

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and combined gas laws

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Gases Class SHORT

ANSWER Answer the
following questions in
the space provided.

Increasing rate of

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effusion. List the following gases in order (Assume all gases are at the same temperature and pressure.) (c) HCl (d) 2. Explain your reasoning for the order of gases you chose in item 1 above. Refer to the kinetic-

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Answers
which precede the
Review Questions.
Chemists try to “see”
the structure of matter
even more closely than
can be seen in any
photograph. ...

negative charge.

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Atomic Theory. 3 – .

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Standing Waves and

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Review for Chemistry
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true or false. Students relate information learned about introductory knowledge gained in chemistry to accurately answer the given questions.

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Gases SECTION 2

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SHORT ANSWER

Answer the following questions in the space provided. 1. State

whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following

circumstances:

increase a.

temperature increases, volume stays the same

decrease b. volume

increases, temperature stays the same

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