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An atoms-first approach to chemistry. Reviewer feedback regarding coverage of advanced topics has been incorporated into the only "atoms first" introductory text available, and informs many of the enhancements to features in MasteringChemistry.; Personalized, interactive learning with MasteringChemistry. Dynamic Study Modules personalize the learning experience for each student.

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Introduction to Atoms Understanding Ideas 1. Name three particles found in an atom. 2. Which two particles are found in an atom's nucleus? 3. An atom has the same number of which two particles? 4. How many protons are in a carbon atom? 5. How are elements identified in terms of their atoms? 6. Explain why scientists use models to study atoms.

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2) All atoms of the same element are exactly alike and have the same mass. Atoms of different elements are different and have different masses. 3) An atom of one element cannot be changed into an atom of a different element by a chemical reaction. 4) Compounds are formed when atoms of more than one element combine in a specific ratio.

Chapter 3-Elements and the Periodic Table Lesson 1 ...

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9 Name Date Class © Pearson Education, Inc., publishing as Pearson Prentice Hall. All rights reserved. ____ 8. According to Boyle's law, as the volume of a gas is made

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Introduction to Matter - Bridgeway

Name Elements and the Periodic Table Date Guided Reading and Study Class Introduction to Atoms (pp. 102-108) This section describes the development of atomic models and the structure of atoms.

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