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Chemistry Matter And Change Chapter 9 Assessment Answers

Chemistry Matter and Change: Chapter 9. covalent bond. endothermic reaction. exothermic reaction. Lewis structure. chemical bond that results from the sharing of valence electro.... chemical reaction or process in which a greater amount of ener.... a chemical reaction or process in which more energy is release....

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The products of this reaction are aqueous calcium chloride, carbon dioxide, and water. (s) + + 142-0 (Q) 21 Chemistry: Matter and Change Chapter 9 Study Guide. Name CHAPTER Date Class STUDY GUIDE Section 9.2 Classifying Chemical Reactions In your textbook, read about synthesis, combustion, decomposition, and replacement reactions.

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Solutions Manual Chemistry: Matter and Change • Chapter 9 145 CHAPTER 9 SOLUTIONS MANUAL 35. Aqueous solutions of potassium iodide and silver nitrate are mixed, forming the precipitate silver iodide. chemical equation: $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$ complete ionic equation: $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

Chemical Reactions

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